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$[\text{Ar}]4s^2 3d^2$; (c) Cr: $[\text{Ar}]4s^1 3d^5$; (d)

Fe: $[\text{Ar}]4s^2 3d^6$; (e) Ru: $[\text{Kr}]5s^2 4d^6$. 3.

(a) La: $[\text{Xe}]6s^2 5d^1$, La $3+$: $[\text{Xe}]$; (b) Sm:

$[\text{Xe}]6s^2 4f^6$, Sm $3+$: $[\text{Xe}]4f^5$; (c) Lu:

$[\text{Xe}]6s^2 4f^{14} 5d^1$, Lu $3+$: $[\text{Xe}]4f^{14}$. 5.

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Al is used because it is the strongest reducing agent and the only option listed that can provide sufficient driving force to convert La (III) into La.

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detailed step by step explanations. ... A sample of mass 5 g of an unknown gas produced a pressure of 19.4 kPa in a 10 L ...

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jit.mrooms.net/mod/quiz/review.php?q=82
08&attempt=292403 2/8 37 4 Marks: 1
Answer: 28 What is the number of
electrons? Correct Marks for this
submission: 1/1. 5 Marks: 1 Answer: 30
What is the atomic number? Correct
Marks for this submission: 1/1. 6 Marks: 1

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