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Solution: $x^3 \hat{e}_1 + 3x^2 \hat{e}_2 + 2x \hat{e}_3 = F e_1 + F e_2 + F e_3$ Forces $F e_1$ and $F e_2$ are equal in magnitude, with $F e_1$ pointing along 45 above the x axis and $F e_2$ pointing along 45 below the x axis. The \hat{y} components cancel. Hence, $F e_3$ is along $+\hat{x}$. Fawwaz T. Ulaby and Umberto Ravaioli, Fundamentals of Applied Electromagnetics c 2019 Prentice Hall

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Solution: Two-wire air line: Because medium between wires is air, $\epsilon = \epsilon_0$, $\mu = \mu_0$ and $s = 0$. $d = 2 \text{ cm}$; $a = 1 \text{ mm}$; $s \ll R$ $s = p \text{ fm}$ $c \ll 1 = 2 = 0$ $R_0 = 0$ $L_0 = m_0 p \ln \frac{2d}{2a + s} = 4\pi \times 10^{-7} \text{ p} \ln \frac{2 \times 20}{2 + 2} = 4 \times 10^{-7} \ln[10 + p/99] = 1.2 \text{ (mH/m)}$; $G_0 = 0$ because $s = 0$ $C_0 = \frac{2\pi\epsilon_0}{\ln \frac{2d}{2a + s}} = 9.29 \text{ (pF/m)}$; Fawwaz T. Ulaby and Umberto Ravaioli, Fundamentals of Applied Electromagnetics

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between, it is force of electromagnetism that rules. At the atomic scale, electromagnetism (admittedly in conjunction with some basic quantum effects) governs the interactions between atoms and molecules. It is the force that underlies the periodic table of elements, giving rise to all of chemistry and, through this, much of biology.

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Solution: (d) $\mathbf{E} = 3 + j2 = 3.61e^{j146.31}$; $i(t) = \text{Re}\{3.61e^{j146.31} e^{j\omega t}\} = 3.61 \cos(\omega t + 146.31)$ A: Fawwaz T. Ulaby, Eric Michielssen, and Umberto Ravaioli, Fundamentals of Applied Electromagnetics c 2010 Prentice Hall

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Solution: (a) From Example 1-4, $I = V / Z = 150 / (400 + j105 \times 3 \times 10^{-3}) = 400 / (150 + j315) = 0.3 \angle -36.9$ (A). (b) $i(t) = \text{Re}\{I e^{j\omega t}\} = \text{Re}\{0.3 e^{j\omega t} \angle -36.9}\} = 0.3 \cos(\omega t - 36.9)$ (A). Exercise 1.10 A phasor voltage is given by $V = j5$ V. Find $v(t)$. Solution:

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